

Stereocontrolled Synthesis of the Northern Part of Potent Proteasome Inhibitor TMC-95A

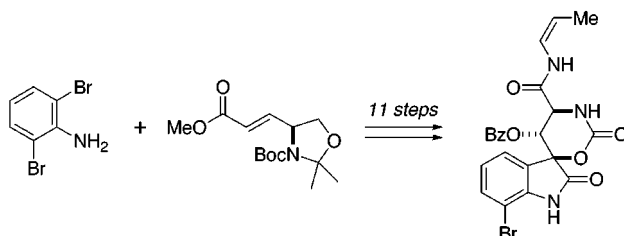
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ABSTRACT

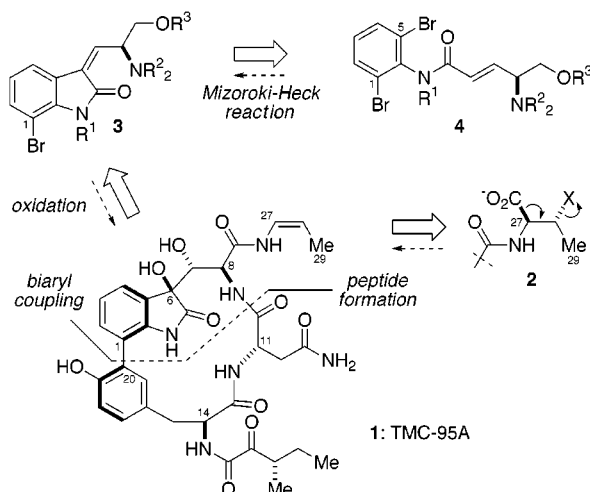


A protected version of the northern part of TMC-95A, a potent and selective proteasome inhibitor, was synthesized with full stereochemical control. Highlights of this synthesis include (i) a (*Z*)-selective Mizoroki–Heck reaction to construct the oxyindole portion, (ii) a diastereoselective epoxidation, (iii) a 6-endo selective epoxide opening by Boc carbonyl group to establish the stereochemistry of C6, and (iv) a 1,3-elimination reaction of the *L*-*allo*-threonine derivative under Mitsunobu conditions to afford the (*Z*)-1-propenylamine.

TMC-95A (**1**, Scheme 1) has been isolated from the fermentation broth of *Apiospora montagnei* Sacc. TC 1093,

which was derived from soil samples.¹ Despite the structural dissimilarity between **1** and other known inhibitors such as lactacystin,² epoxomicin,³ other TMC series,⁴ and peptide-derived molecules,⁵ studies have shown that **1** can specifically inhibit 20S proteasome,^{1b} the catalytic core of proteasome.⁶ Thus, **1** may provide a unique insight into

Scheme 1. Plan for Synthesis of TMC-95A



(1) (a) Kohno, J.; Koguchi, Y.; Nishio, M.; Nakao, K.; Kuroda, M.; Shimizu, R.; Ohnuki, T.; Komatsubara, S. *J. Org. Chem.* **2000**, *65*, 990. (b) Koguchi, Y.; Kohno, J.; Nishio, M.; Takahashi, K.; Okuda, T.; Ohnuki, T.; Komatsubara, S. *J. Antibiot.* **2000**, *53*, 105.

(2) (a) Omura, S.; Fujimoto, T.; Otoguro, K.; Matsuzaki, K.; Moriguchi, R.; Tanaka, H.; Sasaki, Y. *J. Antibiot.* **1991**, *44*, 113. (b) Fenteany, G.; Schreiber, S. L. *J. Biol. Chem.* **1998**, *273*, 8545 and references therein.

(3) (a) Hanada, M.; Sugawara, K.; Kaneta, K.; Toda, S.; Nishiyama, Y.; Tomita, K.; Yamamoto, H.; Konishi, M.; Oki, T. *J. Antibiot.* **1992**, *45*, 1746. (b) Groll, M.; Kim, K. B.; Kairies, N.; Huber, R.; Crews, C. M. *J. Am. Chem. Soc.* **2000**, *122*, 1237 and references therein.

(4) TMC-86A, B and TMC-96: (a) Koguchi, Y.; Kohno, J.; Suzuki, S.-i.; Nishio, M.; Takahashi, K.; Ohnuki, T.; Komatsubara, S. *J. Antibiot.* **1999**, *52*, 1069. (b) Koguchi, Y.; Kohno, J.; Suzuki, S.-i.; Nishio, M.; Takahashi, K.; Ohnuki, T.; Komatsubara, S. *J. Antibiot.* **2000**, *53*, 63. TMC-89A and B: (c) Koguchi, Y.; Nishio, M.; Suzuki, S.-i.; Takahashi, K.; Ohnuki, T.; Komatsubara, S. *J. Antibiot.* **2000**, *53*, 967.

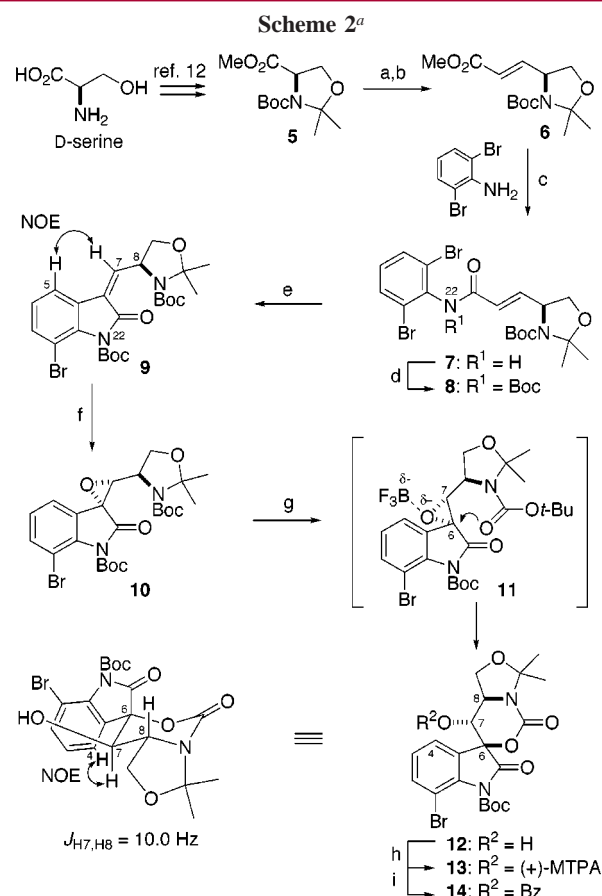
(5) For a recent review on proteasome inhibitors, see: Myung, J.; Kim, K. B.; Crews, C. M. *Med. Res. Rev.* **2001**, *21*, 245.

(6) For reviews on the biology of proteasomes, see: (a) Tanaka, K. *J. Biochem.* **1998**, *123*, 195. (b) Ciechanover, A. *EMBO J.* **1998**, *17*, 7151.

understanding the detailed biological function of proteasomes. Furthermore, proteasomes play important roles in the activation of NF- κ B⁷ and the processing of histocompatibility complex (MHC) class I ligands,⁸ both of which are involved in mounting inflammatory immune responses, thus suggesting the use of TMC-95A and its analogues as potential antiinflammatory agents for treating autoimmune diseases. Notable structural features of **1** are the axially chiral biaryl group, the highly oxidized tryptophan, and the (*Z*)-1-propenylamine in the macrocyclic matrix. The biological importance, as well as the unique architecture, prompted us to undertake the total synthesis of **1**.⁹ Here we report the stereocontrolled synthesis of a protected version of the northern part of **1**.

As outlined in Scheme 1, the structure of **1** can be retrosynthetically divided into two fragments that could be reassembled to the macrocyclic structure by a peptide bond formation and a Pd(0)-catalyzed biaryl coupling.^{9a-c,10} The northern part can be further dissected so that the (*Z*)-1-propenylamine portion could be prepared by the decarboxylative 1,3-elimination of carboxylate **2**. The oxidized tryptophan moiety could be synthesized by the oxidation of **3**, which in turn could be prepared by an intramolecular Mizoroki–Heck reaction of dibromide **4**.¹¹

As shown in Scheme 2, DIBAL reduction followed by Wittig reaction of a known ester **5**, which was derived from D-serine,¹² afforded α,β -unsaturated ester **6** in 71% overall yield.¹³ Amidation of the ester **6** with 2,6-dibromoaniline in the presence of Me₃Al resulted in **7** (59% yield),¹⁴ followed by protection of the N22-amide with a Boc group to produce **8** in 88% yield. After numerous conditions were screened, intramolecular Mizoroki–Heck reaction of **8** was reliably and reproducibly carried out at room temperature,¹⁵ employing “ligandless” conditions¹⁶ that involved a catalytic amount of Pd₂(dba)₃ in the presence of Et₃N, to afford the trisubstituted olefin **9** in 86% yield (*Z*:*E* > 20:1).¹⁷



^a (a) DIBAL, toluene, -78°C ; (b) Ph₃P=CHCO₂Me, CH₂Cl₂, 71% (2 steps); (c) 2,6-dibromoaniline, Me₃Al, toluene; then **6**, 0°C to rt, 59%; (d) (Boc)₂O, Et₃N, DMAP, CH₂Cl₂, rt, 88%; (e) Pd₂(dba)₃·CHCl₃ (0.2 equiv), Et₃N, THF–NMP (1:1), rt, 86%; (f) DMDO, CH₂Cl₂, rt; (g) BF₃·Et₂O (1.5 equiv), CH₂Cl₂, -78 to 0°C , 87% (2 steps); (h) (+)-MTPACl, Et₃N, DMAP, CH₂Cl₂, rt, 98%; (i) BzCl, Et₃N, DMAP, CH₂Cl₂, rt, 83%.

(7) Palombella, V. J.; Rando, O. J.; Goldberg, A. L.; Maniatis, T. *Cell* **1994**, *78*, 773.

(8) Rock, K. L.; Gramm, C.; Rothstein, L.; Clark, K.; Stein, R.; Dick, L.; Hwang, D.; Goldberg, A. L. *Cell* **1994**, *78*, 761.

(9) For synthetic studies on TMC-95 from other laboratories, see: (a) Lin, S.; Danishefsky, S. J. *Angew. Chem., Int. Ed.* **2001**, *40*, 1967. (b) Albrecht, B. K.; Williams, R. M. *Tetrahedron Lett.* **2001**, *42*, 2755. (c) Ma, D.; Wu, Q. *Tetrahedron Lett.* **2001**, *42*, 5279. (d) Ma, D.; Wu, Q. *Tetrahedron Lett.* **2000**, *41*, 9089.

(10) For recent reviews on biaryl synthesis, see: (a) Lloyd-Williams, P.; Giralt, E. *Chem. Soc. Rev.* **2001**, *30*, 145. (b) Stanforth, S. P. *Tetrahedron* **1998**, *54*, 263.

(11) For excellent reviews on Mizoroki–Heck reactions, see: (a) Link, J. T.; Overman, L. E. *Intramolecular Heck Reactions in Natural Product Chemistry*. In *Metal-catalyzed Cross-coupling Reactions*; Diedrich, F., Stang, P. J., Eds.; Wiley-VCH Verlag GmbH: Weinheim, 1998; pp 231–269. (b) Beletskaya, I. P.; Cheprakov, A. V. *Chem. Rev.* **2000**, *100*, 3009. (c) de Meijere, A.; Meyer, F. E. *Angew. Chem., Int. Ed. Engl.* **1994**, *33*, 2379.

(12) (a) Garner, P.; Park, J. M. *J. Org. Chem.* **1987**, *52*, 2361. (b) Garner, P.; Park, J. M. *Org. Synth.* **1992**, *70*, 18.

(13) The numbering of compounds in this paper corresponds to that of TMC-95A.

(14) (a) Lipton, M. F.; Basha, A.; Weinreb, S. M. *Org. Synth.* **1980**, *59*, 49. (b) Overman, L. E.; Paone, D. V.; Stearns, B. A. *J. Am. Chem. Soc.* **1999**, *121*, 7702.

(15) Reaction temperature is crucial for the regioselectivity of this cyclization. For instance, the *E/Z* ratio of **9** was decreased to 3/1 when the reaction was conducted at 50°C .

(16) Madin, A.; Overman, L. E. *Tetrahedron Lett.* **1992**, *33*, 4859.

As shown in Scheme 3, the (*Z*)-selectivity of this facile intramolecular cyclization can be explained by the *syn*-insertion of the alkyl palladium into (*E*)-olefin (**15** \rightarrow **16** and/or **17**), followed by bond rotation (**16** \rightarrow **18** and/or **17** \rightarrow **19**), and *syn*-elimination of the C6-hydride to provide predominantly (*Z*)-oxyindole **9** (Scheme 3). The bulky Boc group protecting the N22-amide of **9** can explain the lack of further Pd(0)-insertion into the C1-bromide of **9**, which would cause undesirable side reactions, and maintain the catalytic cycle.¹⁸

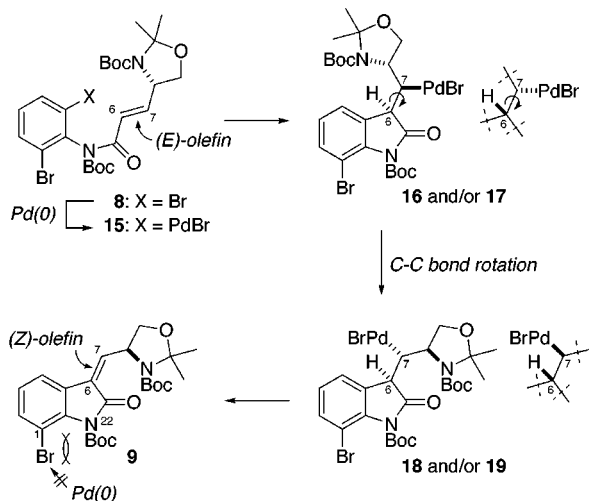
Having secured the route to the oxyindole **9**, the α,β -unsaturated olefin was oxidized using 3,3-dimethyldioxirane (DMDO),¹⁹ to yield epoxide **10** as a single diastereomer (Scheme 2). Interestingly, we discovered that activation of

(17) While this paper was in preparation, we noticed that Lin and Danishefsky (see ref 9a) planned a similar route to this involving Heck cyclization of a closely related substrate.

(18) Macor, J. E.; Ogilvie, R. J.; Wythes, M. J. *Tetrahedron Lett.* **1996**, *37*, 4289.

(19) Danishefsky, S. J.; Bilodeau, M. T. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 1380.

Scheme 3. Proposed Mechanism of (*Z*)-Selective Mizoroki–Heck Reaction



the epoxide of **10** with $\text{BF}_3 \cdot \text{Et}_2\text{O}$ selectively provided a six-membered carbamate **12** with a concomitant loss of the *tert*-butyl group in 87% yield (2 steps).²⁰ The observed selectivity was reasoned that, because of the electronic assistance of the aromatic group, the C6–O bond of **11** would be more susceptible to cleavage than the C7–O bond. The stereostructure of **12** was unambiguously assigned using the coupling constant ($J_{\text{H}7,\text{H}8} = 10.0$ Hz) and the observed NOE between H4 and H7. The results indicate that not only did the epoxidation selectively occur from the opposite side of the C8–NRBoc group of **9**, but the resulting epoxide in **10** was opened via an $\text{S}_{\text{N}}2$ pathway. The enantiomeric excess of **12** was calculated to be 94% after derivatization to its (+)-MTPA ester **13**. For further synthetic manipulations, the secondary alcohol of **12** was protected as its benzoate ester to furnish **14** in 83% yield.

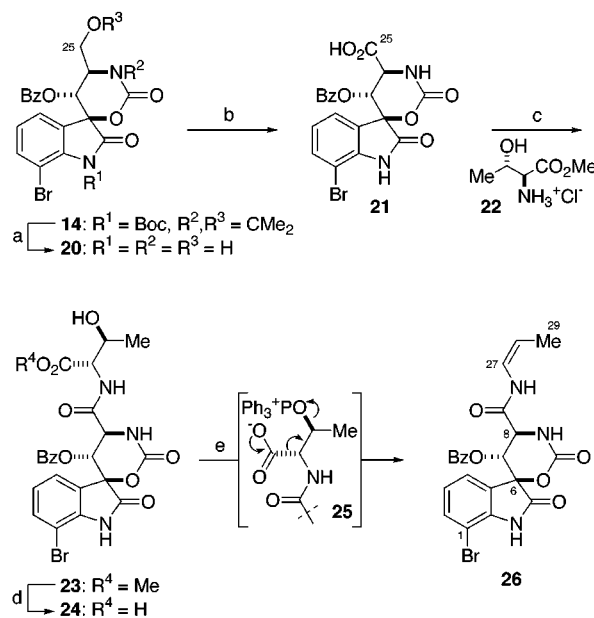
Before building the (*Z*)-1-propenylamine moiety, deprotection and oxidation of the C25-hydroxyl were necessary (Scheme 4). Thus, treatment of **14** in the presence of a catalytic amount of TfOH in $\text{CF}_3\text{CH}_2\text{OH}$ simultaneously removed the acetonide and Boc groups and generated the desired alcohol **20** in 64% yield.²¹ Oxidation of the alcohol **20** using catalytic CrO_3 and H_5IO_6 as a reoxidant provided carboxylic acid **21**,²² which was then coupled with *L*-allo-threonine methyl ester **22** by the action of EDC and HOBt, resulting in amide **23** (60%, 2 steps). Subsequent saponification of methyl ester **23** led to α -alkoxy carboxylic acid **24**. Upon treatment of **24** with Mitsunobu conditions (Ph_3P ,

(20) For a related approach for forming a six-membered carbamate, see: Urabe, H.; Aoyama, Y.; Sato, F. *Tetrahedron* **1992**, *48*, 5639.

(21) (a) Kobayashi, S.; Reddy, R. S.; Sugiura, Y.; Sasaki, D.; Miyagawa, N.; Hiram, M. *J. Am. Chem. Soc.* **2001**, *123*, 2887. (b) Holcombe, J. L.; Livinghouse, T. *J. Org. Chem.* **1986**, *51*, 111.

(22) Zhao, M.; Li, J.; Song, Z.; Desmond, R.; Tschaen, D. M.; Grabowski, E. J. J.; Reider, P. J. *Tetrahedron Lett.* **1998**, *39*, 5323.

Scheme 4^a



^a (a) TfOH (0.5 equiv), $\text{CF}_3\text{CH}_2\text{OH}$, 0 °C to rt, 64%; (b) CrO_3 (0.1 equiv), H_5IO_6 , wet MeCN, 0 °C; (c) **22** (1.5 equiv), EDC, HOBt, NMM, DMF, 0 °C to rt, 60% (2 steps); (d) $\text{LiOH} \cdot \text{H}_2\text{O}$, THF– H_2O (1:1), 0 °C; (e) DEAD, PPh_3 , THF, –78 °C to rt, 40% (2 steps).

DEAD), the decarboxylative 1,3-elimination proceeded smoothly,²³ presumably through **25**, to yield the protected northern part of TMC-95A (**26**) in 40% yield (2 steps).²⁴

In conclusion, we have attained a highly stereocontrolled synthesis of **26** through (i) (*Z*)-selective Mizoroki–Heck reaction of dibromide **8** to construct the oxyindole portion, (ii) a diastereoselective epoxidation procedure (**9** → **10**), (iii) a 6-endo selective epoxide ring-opening reaction to establish the stereochemistry of C6 stereocenter (**10** → **12**), and (iv) a mild 1,3-elimination reaction of *L*-allo-threonine to build the (*Z*)-1-propenylamine (**24** → **26**). Further studies directed toward the total synthesis of TMC-95A, including derivatization of **26** to a suitable form for coupling with a southern part, are currently underway in our laboratory.

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(23) Pansare, S. V.; Vederas, J. C. *J. Org. Chem.* **1989**, *54*, 2311.

(24) Physical data of **26**: $[\alpha]_{\text{D}}^{25} +86.0$ (c 0.62, CHCl_3); IR (film) 3280, 1741, 1672, 1621, 1264, 1107, 708 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3) δ 1.08 (3H, dd, $J = 7.5, 1.5$ Hz, H29), 4.72 (1H, dq, $J = 9.0, 7.5$ Hz, H28), 5.31 (1H, d, $J = 9.5$ Hz, H8), 6.02 (1H, d, $J = 9.5$ Hz, H7), 6.43 (1H, ddd, $J = 10.5, 9.0, 1.5$ Hz, H27), 6.96 (1H, s, amide-NH), 6.99 (1H, dd, $J = 8.5, 7.5$ Hz, H3), 7.36 (1H, d, $J = 8.5$ Hz, H4), 7.44 (2H, dd, $J = 8.0, 7.5$ Hz, Bz), 7.50 (1H, d, $J = 7.5$ Hz, H2), 7.59 (1H, t, $J = 7.5$ Hz, Bz), 7.93 (2H, d, $J = 7.5$ Hz, Bz), 8.83 (1H, d, $J = 10.5$ Hz, amide-NH), 9.57 (1H, s, amide-NH); ^{13}C NMR (125 MHz, CDCl_3) δ 10.7, 52.5, 66.5, 80.7, 103.9, 109.9, 120.8, 124.6, 125.1, 125.2, 127.8, 128.8, 130.0, 134.3, 134.6, 141.2, 151.4, 163.9, 165.2, 173.6; MALDI-TOF MS calcd for $\text{C}_{22}\text{H}_{19}\text{BrN}_3\text{O}_6$ [$\text{M} + \text{H}^+$] 500.0457, found 500.0433.